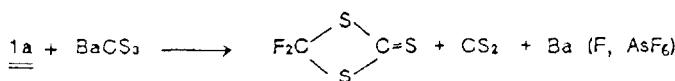
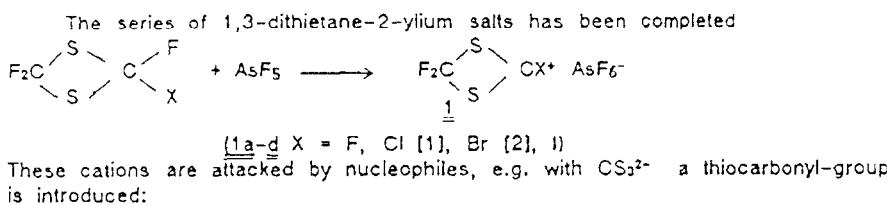
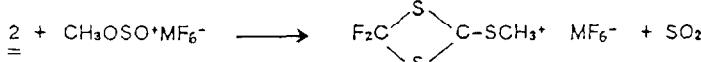


## NEW PERFLUORO-1,3-DITHIETANE-CHEMISTRY

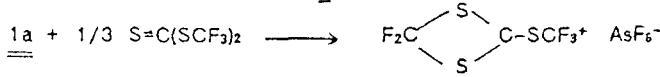
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Leobenerstrasse NW2, D-2800 Bremen 33 (F.R.G.)

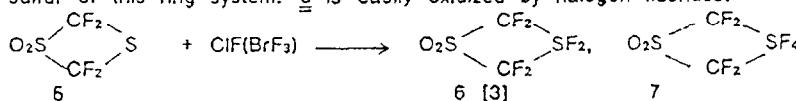
2 is alkylated in liq.  $\text{SO}_2$  to give an trithiocarbenium ion:



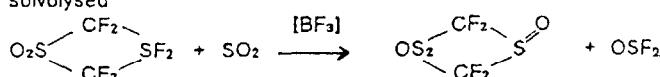
The perfluorinated analog of 3 is obtained in a one step synthesis



A second aspect in the chemistry of dithietanes are oxidative additions to the sulfur of this ring system. 5 is easily oxidized by halogen fluorides:



With  $\text{AsF}_5$  6 forms a stable sulfonium salt, in the presence of  $\text{BF}_3$  6 is readily solvolyzed



The  $\text{BF}_3$ -catalysed solvolysis seems to be a general synthetic method, several other examples will be presented.

Stabilities, structures, spectroscopic properties and reactivities of the compounds described will be discussed.

1 A. Waterfeld, R. Mews, *Chem. Ber.*, **118** (1985) 4997.

2 J. Antel, K. Harms, P.G. Jones, R. Mews, G.M. Sheldrick, A. Waterfeld, *Chem. Ber.*, **118** (1985) 5006.

3 W. Sundermeyer, M. Witz, *J. Fluorine Chem.*, **26** (1984) 395.